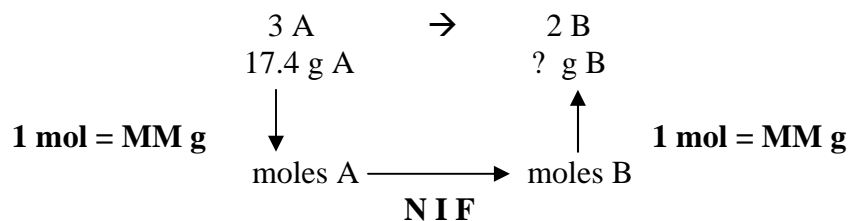
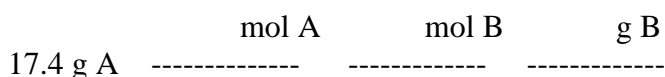


STEPS:

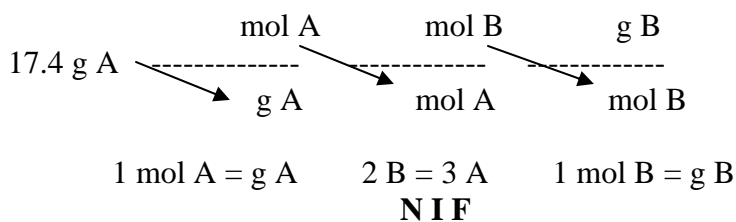
1. Find your starting chemical / grams and your final chemical (eg. 17.4 grams A form how much B?)
2. Make a road map. EX: 17.5 grams A \rightarrow moles A \rightarrow moles B \rightarrow grams B



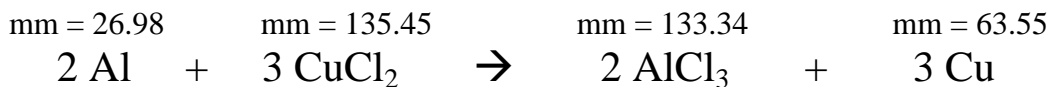
3. Put new words on top of 3 conversion bars, in order:



4. Slide words down to get rid of them:



5. Put appropriate equalities into the conversions.

Use the Following Equation for ALL of the following problems

1. You weigh out 78.5 grams of Aluminum. How many grams of Cu will form? (277 g Cu)
2. You weigh out 175 grams of CuCl_2 . How many grams of AlCl_3 will form? (115 g AlCl_3)
3. You form 92.4 grams of Copper. How many grams of Aluminum were reacted? (26.2 g Al)
4. You form 49.2 grams of AlCl_3 . How many grams of CuCl_2 were reacted? (75.0 g CuCl_2)